## ENANTIOMERIC SYNTHESES OF 6(R), 7(R) AND 6(S), 7(S) trans- AND cis-LAUREDIOL

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**Abstract:** The title compounds have been synthesized by using acetylenic coupling procedures, asymmetric epoxidation and stereo- and regio-selective openings of the epoxides.

From the red algae of the genus <u>Laurencia</u> have been isolated a wide variety of halogenated cyclic ethers characterized by a straight-chain  $C_{15}$ atom skeleton and a terminal enyne function<sup>1</sup>). Related compounds have also been found in <u>Aplysia<sup>1</sup></u>) probably having a dietary origin in <u>Laurencia</u>. Irie and his coworkers<sup>2</sup>) reported the isolation from <u>Laurencia nipponica</u> Yamada of optical isomers with 6(S),7(S) and 6(R),7(R) configurations of 6,7-dihydroxy-3,12-<u>trans</u>-9-<u>cis</u>-pentadeca-3,9,12-trien-1-yne (<u>trans</u>-laurediol) (<u>1</u>,<u>2</u>) and 6,7-dihydroxy-3,9-<u>cis</u>-12-<u>trans</u>-pentadeca-3,9,12-trien-1-yne (<u>cis</u>-laurediol) (<u>3</u>,<u>4</u>) as well as their corresponding acetates (<u>5</u>--><u>8</u>).



These compounds have been repeatedly proposed as biosynthetic precursors of the cyclic ethers<sup>1</sup>) by electrophilic cyclization<sup>2</sup>). In this communication we report the total syntheses<sup>3</sup>) of these metabolites in their enantiomeric natural form by a general methodology based on acetylenic coupling catalyzed by copper<sup>4</sup>), asymmetric epoxidation<sup>5</sup>) and regio- and stereo-selective openings of the epoxides<sup>6</sup>.

For the synthesis of <u>1</u> and its acetate <u>5</u> a dienynol <u>17</u> was prepared from ethyl bromide and propargyl-O-THP ether by acetylenic coupling catalyzed by copper (I) chloride<sup>4)</sup> (Scheme <u>1</u>)<sup>7)</sup>.

<u>17</u> was selectively hydrogenated using Lindlar's catalyst<sup>8</sup>) in 87% yield and submitted to asymmetric epoxidation under the standard conditions<sup>5</sup>),





Ti(OPr<sup>i</sup>)<sub>4</sub>, D-(+)-diethyl tartrate, tert-butyl hydroperoxide, CH<sub>2</sub>Cl<sub>2</sub>, -20<sup>o</sup>C, 20 hrs., giving the epoxyalcohol <u>19</u> in 82% yield with over 95% ee<sup>9</sup>),  $|\alpha|_D^{25}$ -10.2<sup>o</sup> (c 3.45, CHCl<sub>3</sub>) (Scheme II).



At this point of the synthesis, we need the isomerization of the 2,3to the <u>threo</u>-1,2-epoxyalcohol to create the right configuration of the diol system by a carbon nucleophile attack to the terminal epoxide at the C-1 position. The described procedure<sup>6</sup>) using titanium (IV) isopropoxide assisted opening of chiral 2,3-epoxyalcohols from <u>E</u>-allylic alcohols has proved to be an excellent way to obtain <u>21</u> in 65% yield from <u>19</u>. The <u>threo</u>-1,2-epoxide <u>21</u>,  $|\alpha|_D^{25}$ -10.7° (c 2.09, ether), was protected with ethyl vinyl ether, in CH<sub>2</sub>Cl<sub>2</sub>, using pyridium <u>p</u>-toluenesulphonate as catalyst<sup>10</sup>) and was treated with the lithium salt of propargyl-O-THP ether (1.5 equiv.) and BF<sub>3</sub>.OEt<sub>2</sub><sup>11</sup>)(1.5 equiv.) at -78°C for 0.5 hr. to give <u>23</u> in 73% yield (Scheme <u>III</u>).

The acidic deprotection of <u>23</u> in methanol, protection of <u>24</u> with 2-methoxypropene in CH<sub>2</sub>Cl<sub>2</sub> and deprotection of some <u>25</u> formed with methanol and a catalytic amount of acetic acid yielded <u>26</u>, 85% yield,  $|\alpha|_{D}^{25}$ -25.4° (c 1.65, CHCl<sub>3</sub>). Reduction of the yne-function with lithium aluminum hydride<sup>12</sup>) in THF at 0°C for 2 hrs. led to <u>27</u> in 93% yield  $|\alpha|_{D}^{25}$ -25.4° (c 1.65, CHCl<sub>3</sub>). This compound was oxidised with excess of manganese dioxide in CH<sub>2</sub>Cl<sub>2</sub> at room temperature and treated without purification with triphenyl phosphine (5.5 equiv.) and carbon tetrabromide (2.7 equiv.) at 0°C in CH<sub>2</sub>Cl<sub>2</sub> for 2 hrs. to yield <u>29</u>,  $|\alpha|_{D}^{25}$ -27.9° (c 2.40, CHCl<sub>3</sub>), in 73% overall yield <sup>13</sup>). Treatment in ether for 10 min. with 2.1 equiv. of n-BuLi<sup>14</sup>) gave <u>30</u> in 82% yield,  $|\alpha|_D^{25}$ -25.0° (c 1.32, CHCl<sub>3</sub>).



The deprotection of <u>30</u> with methanol and a catalytic amount of <u>p</u>-toluenesulphonic acid gave <u>1</u>,  $|\alpha|_D^{25}$ -14.7° (c 1.2, CCl<sub>4</sub>) in 85% yield. Acetylation under standard conditions (Ac<sub>2</sub>O, Py) gave the diacetate <u>5</u>,  $|\alpha|_D^{25}$ +5.3° (c 0.7, CHCl<sub>3</sub>) in 95% yield. According to the optical purity observed in the epoxide <u>25</u> these products must have more than 95% ee<sup>14</sup>.

The syntheses of the geometrical isomer <u>cis</u>-laurediol <u>3</u>,  $|\alpha|_D^{25} - 9.3^\circ$ (c 0.9, CHCl<sub>3</sub>)<sup>14</sup>) and its corresponding acetate <u>7</u>,  $|\alpha|_D^{25} + 3.5^\circ$  (c 0.8, CHCl<sub>3</sub>), were performed in a totally similar manner, but using Lindlar's hydrogenation of <u>26</u> to get the <u>Z</u>-olefin<sup>15</sup>). The syntheses of isomers <u>2,6</u> and <u>4,8</u> were carried out following a similar procedure but changing the isomer of the tartaric acid ester used at the asymmetric epoxidation step.

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References and notes:

 a) Moore, R.E.; <u>Algal Nonisoprenoids in Marine Natural Products</u>, Ed. by P.J. Scheuer, N.Y., 1978, <u>Vol I</u>, 44-121.

b) Faulkner, D.J.; <u>Natural Products Report</u>, 1984, <u>1</u>, 251-280.

- 2) Kurosawa, E.; Fukuzawa, A., Irie, T.; <u>Tet. Lett.</u>, 1972, <u>21</u>, 2121.
- 3) A total synthesis of 6(S), 7(S) <u>trans</u>-laurediol <u>1</u> has been recently reported: Fukuzawa, A.; Sato, H.; Miyamoto, M.; Masamune, T.; <u>Tet. Lett.</u>, 1986, <u>27</u>, 2901.
- Taginachi, H.; Mathai, I.M.; Miller, S.I.; Org. Synth., 1970, <u>50</u>, 97.
- a) Katsuki, T.; Sharpless, K.B.; <u>J. Am. Chem. Soc.</u>, 1980, <u>102</u>, 5974.

b) Martín, V.S., Woodard, S.S.; Katsuki, T.; Yamada, Y.; Ikeda, M.; Sharpless, K.B.; <u>J. Am. Chem. Soc.</u>, 1981, <u>103</u>, 6237.
c) Hill, P.G.; Rossiter, B.E.; Sharpless, B.B.; <u>J. Org. Chem.</u>, 1983, <u>48</u>, 3603.

- 6) See the preceding paper.
- 7) a) i)n-BuLi, THF, HMPA, -78°C, 1 hr.; ii)EtBr, -78°C-->R.T., 14 hrs.; iii)HCl (conc.)(cat.), MeOH, R.T., 10 hrs., 92% overall yield; b) i)LiAlH<sub>4</sub>, THF, reflux, 2 hrs. 87%; ii)Br<sub>3</sub>P, pentane, 0°C, 0.5 hr., 78%; c) i)BrMgC ≡ CCH<sub>2</sub>OTHP, THF, Cu<sub>2</sub>Cl<sub>2</sub> (cat.), 0°C-->reflux, 4 hrs.; ii)(C<sub>6</sub>H<sub>5</sub>)<sub>3</sub>PBr<sup>+</sup>Br<sup>-</sup>, CH<sub>2</sub>Cl<sub>2</sub>, R.T., 74%; d) BrMgC ≡ CCH<sub>2</sub>OTHP, THF, Cu<sub>2</sub>Cl<sub>2</sub> (cat.), Me<sub>2</sub>S, 0°C-->reflux, 4 hrs., 83%; e) i)HCl (conc)(cat.), MeOH, R.T., 5 hrs.; ii)LiAlH<sub>4</sub>, ether, R.T., 14 hrs., 81% overall yield.
- <u>17</u> must be carefully purified in order to obtain a good rate in Lindlar's hydrogenation.
- 9) The optical purities of all the epoxides were checked by using NMR on the epoxyalcohol acetates with  $Eu(hfbc)_3$  as chiral shift reagent and/or the Mosher's ester.
- 10) Any attempt to protect the hydroxy group under basic conditions gave a substantial amount of isomerization to the 2,3-epoxyalcohol.
- 11) Yamaguchi, M.; Hirao, I.; J. Chem. Soc. Chem. Comm., 1984, 202.
- 12) Magoon, E.F.; Slaugh, L.H.; Tetrahedron, 1967, 23, 4509.
- 13) Corey, E.J.; Lansburg, P.T.; Cashman, J.R.; Kantner, S.S.; <u>J. Am. Chem.</u> <u>Soc.</u>, 1984, <u>106</u>, 1501.
- 14) The reported optical rotation for 6(R), 7(R) <u>trans</u>-laurediol <u>2</u> obtained from laurencin by Zn-AcOH-EtOH treatment is  $|\alpha|_D + 27.2^{\circ}$ . In a similar manner the value for 6(S), 7(S) <u>cis</u>-laurediol is  $|\alpha|_D - 19.5^{\circ}$ . See Ref. 2.
- 15) Corey, E.J.; Ruden, R.A.; <u>Tet. Lett.</u>, 1973, 1495. By this procedure a ratio of 1:1 of <u>30</u> and the <u>cis</u> isomer was reached. Both products were separated by column chromatography.

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